

FORM PTO-1390 (Modified)  
(REV 11-2000)

U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY'S DOCKET NUMBER

TRANSMITTAL LETTER TO THE UNITED STATES

45503-00011

DESIGNATED/ELECTED OFFICE (DO/EO/US)

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR

CONCERNING A FILING UNDER 35 U.S.C. 371

09/937154

INTERNATIONAL APPLICATION NO.  
PCT/DE00/00872

INTERNATIONAL FILING DATE  
22 March 2000

PRIORITY DATE CLAIMED  
24 March 1999

TITLE OF INVENTION

CHEMICALLY PASSIVATED OBJECT MADE OF MAGNESIUM OR ALLOYS THEREOF

APPLICANT(S) FOR DO/EO/US

KURZE, Peter; KRUGER, Ulrike; KOHLER, Marco; BANERJEE, Dora

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☐ This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (24) indicated below.
4. ☒ The US has been elected by the expiration of 19 months from the priority date (Article 31).
5. ☒ A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
  - a. ☒ is attached hereto (required only if not communicated by the International Bureau).
  - b. ☐ has been communicated by the International Bureau.
  - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US).
6. ☒ An English language translation of the International Application as filed (35 U.S.C. 371(c)(2)).
  - a. ☒ is attached hereto.
  - b. ☐ has been previously submitted under 35 U.S.C. 154(d)(4).
7. ☐ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
  - a. ☐ are attached hereto (required only if not communicated by the International Bureau).
  - b. ☐ have been communicated by the International Bureau.
  - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
  - d. ☐ have not been made and will not be made.
8. ☐ An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
9. ☒ An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
10. ☐ An English language translation of the annexes of the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).
11. ☐ A copy of the International Preliminary Examination Report (PCT/IPEA/409).
12. ☐ A copy of the International Search Report (PCT/ISA/210).

Items 13 to 20 below concern document(s) or information included:

13. ☐ An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
14. ☒ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
15. ☒ A **FIRST** preliminary amendment.
16. ☐ A **SECOND** or **SUBSEQUENT** preliminary amendment.
17. ☐ A substitute specification.
18. ☐ A change of power of attorney and/or address letter.
19. ☐ A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
20. ☐ A second copy of the published international application under 35 U.S.C. 154(d)(4).
21. ☐ A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
22. ☐ Certificate of Mailing by Express Mail
23. ☐ Other items or information:

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR 1.53) <b>09/937154</b>	INTERNATIONAL APPLICATION NO. <b>PCT/DE00/00872</b>	ATTORNEY'S DOCKET NUMBER <b>45503-00011</b>
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24. The following fees are submitted:				CALCULATIONS PTO USE ONLY																					
<b>BASIC NATIONAL FEE ( 37 CFR 1.492 (a) (1) - (5) ) :</b> <input type="checkbox"/> Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO ..... <b>\$1000.00</b> <input checked="" type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO ..... <b>\$860.00</b> <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO ..... <b>\$710.00</b> <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4) ..... <b>\$690.00</b> <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) ..... <b>\$100.00</b> <b>ENTER APPROPRIATE BASIC FEE AMOUNT =</b>																									
Surcharge of <b>\$130.00</b> for furnishing the oath or declaration later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492 (e)).																									
<table border="1"> <thead> <tr> <th>CLAIMS</th> <th>NUMBER FILED</th> <th>NUMBER EXTRA</th> <th>RATE</th> </tr> </thead> <tbody> <tr> <td>Total claims</td> <td>18 - 20 =</td> <td>0</td> <td>x \$18.00</td> </tr> <tr> <td>Independent claims</td> <td>2 - 3 =</td> <td>0</td> <td>x \$80.00</td> </tr> <tr> <td colspan="3">Multiple Dependent Claims (check if applicable).</td> <td><input type="checkbox"/></td> </tr> <tr> <td colspan="3"><b>TOTAL OF ABOVE CALCULATIONS</b></td> <td><b>=</b></td> </tr> </tbody> </table>				CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE	Total claims	18 - 20 =	0	x \$18.00	Independent claims	2 - 3 =	0	x \$80.00	Multiple Dependent Claims (check if applicable).			<input type="checkbox"/>	<b>TOTAL OF ABOVE CALCULATIONS</b>			<b>=</b>		
CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE																						
Total claims	18 - 20 =	0	x \$18.00																						
Independent claims	2 - 3 =	0	x \$80.00																						
Multiple Dependent Claims (check if applicable).			<input type="checkbox"/>																						
<b>TOTAL OF ABOVE CALCULATIONS</b>			<b>=</b>																						
<input checked="" type="checkbox"/> Applicant claims small entity status. (See 37 CFR 1.27). The fees indicated above are reduced by 1/2.																									
<b>SUBTOTAL</b>				<b>= \$430.00</b>																					
Processing fee of <b>\$130.00</b> for furnishing the English translation later than <input type="checkbox"/> 20 <input type="checkbox"/> 30 months from the earliest claimed priority date (37 CFR 1.492 (f)).																									
<b>TOTAL NATIONAL FEE</b>				<b>= \$430.00</b>																					
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable).				<input checked="" type="checkbox"/>																					
<b>TOTAL FEES ENCLOSED</b>				<b>= \$470.00</b>																					
				Amount to be: refunded	\$																				
				charged	\$																				

- a. ☒ A check in the amount of **\$470.00** to cover the above fees is enclosed.
- b. ☐ Please charge my Deposit Account No. \_\_\_\_\_ in the amount of \_\_\_\_\_ to cover the above fees. A duplicate copy of this sheet is enclosed.
- c. ☒ The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. **22-0585**. A duplicate copy of this sheet is enclosed.
- d. ☐ Fees are to be charged to a credit card. **WARNING:** Information on this form may become public. **Credit card information should not be included on this form.** Provide credit card information and authorization on PTO-2038.

**NOTE:** Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO:

Thomas R. Boland, Esquire  
 Vorys, Sater, Seymour and Pease, LLP  
 1828 L Street, NW  
 Suite 1111  
 Washington, DC 20036

SIGNATURE

Thomas R. Boland

NAME

24,093

REGISTRATION NUMBER

September 21, 2001

DATE

09/937154

JCO3 Rec'd PCT/PTO 21 SEP 2001

PATENT APPLICATION

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of:

KRUZE, Peter; KRUGER, Ulrike;  
KOHLE, Marco; BANERJEE, Dora

Group Art Unit:

International Application No.: PCT/DE00/00872

Examiner:

Filed: 22 March 2000

For: CHEMICALLY PASSIVATED OBJECT MADE OF MAGNESIUM OR ALLOYS  
THEREOF

PRELIMINARY AMENDMENT

Commissioner for Patents  
Washington, D.C. 20231

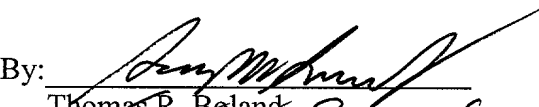
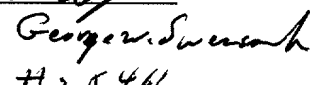
Sir:

In advanced prosecution and before calculating the filing fees, please amend the above-identified application as follows:

IN THE CLAIMS:

Please replace all original claims with amended claims [A]. A marked-up version of the amended claims is attached as [B].

Respectfully submitted,

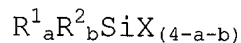
By:   
Thomas R. Boland  
Reg. No. 24,093  
  
#25,461

Vorys, Sater, Seymour  
and Pease LLP  
1828 L Street, NW, Suite 1111  
Washington, D.C. 20036-5109  
Tel: (202) 467-8880

September 21, 2001

204/99003  
WO 22 March 2000

1. Article made of magnesium or its alloys, some or all of whose surface has a conversion coating, characterized in that the conversion coating comprises  $MgO$ ,  $Mn_2O_3$  and  $MnO_2$  plus at least one oxide from the group consisting of vanadium, molybdenum and tungsten.
2. (amended) Article according to Claim 1, characterized in that the conversion coating is obtainable by passivating the article using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
3. (amended) Article according to Claim 1, characterized in that in addition to the conversion coating a polymer coating has been applied which is obtainable by polymerizing and/or crosslinking a solution comprising at least one alkoxysilane compound.
4. Article according to Claim 3, characterized in that the alkoxysilane compound is of the general formula



- X is an alkoxy, aryloxy or acyloxy group of 1 to 12 carbon atoms, preferably of 1 to 4 carbon atoms, and in

particular is selected from the group consisting of methoxy, ethoxy, n-propoxy, i-propoxy, butoxy, phenoxy, acetoxy and propionyloxy groups;

- $R^1$  and  $R^2$ , which are identical to or different from one another, are selected from the group consisting of
  - amino, monoalkylamino or dialkylamino radicals;
  - alkyl radicals, especially the alkyl radicals of 1 to 6 carbon atoms, preferably the methyl, ethyl, n-propyl, isopropyl, n-butyl, s-butyl, t-butyl, pentyl, hexyl or cyclohexyl radicals;
  - alkenyl radicals, especially the alkenyl radicals of 2 to 6 carbon atoms, preferably the vinyl, 1-propenyl, 2-propenyl or butenyl radicals;
  - alkynyl radicals, especially the alkynyl radicals of 2 to 6 carbon atoms, preferably the acetylenyl or propargyl radicals;
  - aryl radicals, especially the aryl radicals of 6 to 10 carbon atoms, preferably phenyl or naphthyl radicals;
  - epoxy radicals, especially the epoxy radicals of 3 to 16 carbon atoms, preferably the glycidyl, glycidyl ether, glycidyl ester or glycidyloxyalkyl radicals; or group X described above; and

- a and b, which are identical to or different from one another, are 0, 1, 2 or 3, the sum of a and b not exceeding 3.
5. Article according to Claim 4, characterized in that the alkoxyasilane compound is a tetraalkoxyasilane, epoxyalkoxyasilane or aminoalkoxyasilane.
  6. Article according to Claim 5, characterized in that the alkoxyasilane compound is selected from the group consisting of tetraethoxyasilane, 3-glycidyloxypropyltrimethoxyasilane, 3-aminopropyltrimethoxyasilane and 3-(aminoethylamino) propyltrimethoxyasilane.
  7. (amended) Article according to Claim 3, characterized in that the solution additionally comprises a compound capable of forming a titanium complex.
  8. Article according to Claim 7, characterized in that the compound capable of forming a titanium complex is an alkoxytitanium compound, a titanate ester or a titanium chelate and in particular is of the formula  $Ti(OR)_4$  in which R is an alkyl radical of 1 to 6 carbon atoms selected preferably from the group consisting of methyl, ethyl, n-propyl, isopropyl and butyl radicals.
  9. Article according to Claim 8, characterized in that the compound capable of forming a titanium complex is tetraethoxytitanate  $Ti(OC_2H_5)_4$ .
  10. (amended) Article according to Claim 3, characterized in that the solution additionally comprises at least one dye

which is soluble in a polar solvent, in particular a metal complex dye.

11. Process for producing a conversion coating on an article made of magnesium or its alloys, characterized in that the article is subjected to passivation using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
12. Process according to Claim 11, characterized in that the passivation is conducted within a pH range of the aqueous passivating electrolyte of from 7.0 to 8.0.
13. (amended) Process according to Claim 11, characterized in that the passivation is conducted at a temperature of the aqueous passivating electrolyte of from 15 to 50 °C, in particular from 20 to 30 °C.
14. (amended) Process according to Claim 11, characterized in that the passivation is conducted for a period of from 2 to 10 minutes.
15. (amended) Process according to Claim 11, characterized in that the concentration of potassium permanganate in the aqueous passivating electrolyte is from 1 to 10 g/l.
16. (amended) Process according to Claim 11, characterized in that the concentration of the alkali metal salt or ammonium salt from the group consisting of vanadate, molybdate and

tungstate in the aqueous passivating electrolyte is from 1 to 10 g/l.

17. (amended) Process according to Claim 11, characterized in that a paint or other surface coating material is or has been applied to the conversion coating.

FOR 2004-03-24

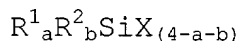


[B]

204/99003  
WO 22 March 2000

# C l a i m s

1. Article made of magnesium or its alloys, some or all of whose surface has a conversion coating, characterized in that the conversion coating comprises  $MgO$ ,  $Mn_2O_3$  and  $MnO_2$  plus at least one oxide from the group consisting of vanadium, molybdenum and tungsten.
2. (amended) Article according to Claim[,]1, characterized in that the conversion coating is obtainable by passivating the article using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
3. (amended) Article according to Claim 1 [or 2], characterized in that in addition to the conversion coating a polymer coating has been applied which is obtainable by polymerizing and/or crosslinking a solution comprising at least one alkoxy silane compound.
4. Article according to Claim 3, characterized in that the alkoxy silane compound is of the general formula



in which

- X is an alkoxy, aryloxy or acyloxy group of 1 to 12 carbon atoms, preferably of 1 to 4 carbon atoms, and in

particular is selected from the group consisting of methoxy, ethoxy, n-propoxy, i-propoxy, butoxy, phenoxy, acetoxy and propionyloxy groups;

- $R^1$  and  $R^2$ , which are identical to or different from one another, are selected from the group consisting of
  - amino, monoalkylamino or dialkylamino radicals;
  - alkyl radicals, especially the alkyl radicals of 1 to 6 carbon atoms, preferably the methyl, ethyl, n-propyl, isopropyl, n-butyl, s-butyl, t-butyl, pentyl, hexyl or cyclohexyl radicals;
  - alkenyl radicals, especially the alkenyl radicals of 2 to 6 carbon atoms, preferably the vinyl, 1-propenyl, 2-propenyl or butenyl radicals;
  - alkynyl radicals, especially the alkynyl radicals of 2 to 6 carbon atoms, preferably the acetylenyl or propargyl radicals;
  - aryl radicals, especially the aryl radicals of 6 to 10 carbon atoms, preferably phenyl or naphthyl radicals;
  - epoxy radicals, especially the epoxy radicals of 3 to 16 carbon atoms, preferably the glycidyl, glycidyl ether, glycidyl ester or glycidyloxyalkyl radicals; or group X described above; and

- a and b, which are identical to or different from one another, are 0, 1, 2 or 3, the sum of a and b not exceeding 3.

5. Article according to Claim 4, characterized in that the alkoxysilane compound is a tetraalkoxysilane, epoxyalkoxysilane or aminoalkoxysilane.

6. Article according to Claim 5, characterized in that the alkoxysilane compound is selected from the group consisting of tetraethoxysilane, 3-glycidyloxypropyltrimethoxysilane, 3-aminopropyltrimethoxysilane and 3-(aminoethylamino) propyltrimethoxysilane.

7. (amended) Article according to Claim [any of Claims] 3 [to 6], characterized in that the solution additionally comprises a compound capable of forming a titanium complex.

8. Article according to Claim 7, characterized in that the compound capable of forming a titanium complex is an alkoxytitanium compound, a titanate ester or a titanium chelate and in particular is of the formula  $Ti(OR)_4$  in which R is an alkyl radical of 1 to 6 carbon atoms selected preferably from the group consisting of methyl, ethyl, n-propyl, isopropyl and butyl radicals.

9. Article according to Claim 8, characterized in that the compound capable of forming a titanium complex is tetraethoxytitanate  $Ti(OC_2H_5)_4$ .

10. (amended) Article according to Claim [any of Claims] 3 [to 9], characterized in that the solution additionally

comprises at least one dye which is soluble in a polar solvent, in particular a metal complex dye.

11. Process for producing a conversion coating on an article made of magnesium or its alloys, characterized in that the article is subjected to passivation using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
12. Process according to Claim 11, characterized in that the passivation is conducted within a pH range of the aqueous passivating electrolyte of from 7.0 to 8.0.
13. (amended) Process according to Claim 11 [or 12], characterized in that the passivation is conducted at a temperature of the aqueous passivating electrolyte of from 15 to 50 °C, in particular from 20 to 30 °C.
14. (amended) Process according to Claim [any of Claims] 11 [to 13], characterized in that the passivation is conducted for a period of from 2 to 10 minutes.
15. (amended) Process according to Claim [any of Claims] 11 [to 14], characterized in that the concentration of potassium permanganate in the aqueous passivating electrolyte is from 1 to 10 g/l.
16. (amended) Process according to Claim [any of Claims] 11 [to 15], characterized in that the concentration of the alkali metal salt or ammonium salt from the group consisting of

vanadate, molybdate and tungstate in the aqueous passivating electrolyte is from 1 to 10 g/l.

17. (amended) Process according to Claim [any of Claims] 11 [to 16], characterized in that a paint or other surface coating material is or has been applied to the conversion coating.

[18. Use of a solution for producing an article according to any of Claims 3 to 10, characterized in that the solution comprises at least one alkoxysilane compound according to any of Claims 3 to 9.]

[19. Use of an article according to any of Claims 1 to 10 and of an article obtainable by a process as claimed in any of Claims 11 to 17 in the motor vehicle industry, electrical and electronics industry, mechanical engineering industry, air travel and space travel.]

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T01250-451250

09/937154

204/99003 WO

22 March 2000

JCO3 Rec'd PGT/ATO 21 SEP 2001

"Chemically passivated article made of  
magnesium or its alloys"

The present invention relates to an article made of magnesium or its alloys which has a conversion coating produced by passivating the surface, to a process for producing such an article, and to its use.

Magnesium and its alloys are the lightest but also least noble metallic construction materials (standard potential of Mg -2.34 volts) and therefore have a very strong tendency to corrode. To counter this disadvantageous property, magnesium and its alloys are treated in aqueous passivating electrolytes. The redox procedure which occurs during this treatment (without an external current source) forms a conversion coating consisting of oxides of the magnesium material and oxide-type reaction products originating from the constituents of the aqueous passivating electrolyte.

The term "conversion coating" refers here and below to a coating which is formed not by application to a surface but rather by chemical transformation (conversion) of the metallic surface and of various constituents of the aqueous passivating electrolyte (cf. H. Simon, M. Thoma "Angewandte Oberflächentechnik für metallische Werkstoffe", Carl Hanser Verlag, Munich (1985) p. 4).

For example, the chromation of articles made of magnesium or its alloys is known. The corresponding processes are described in particular in the MIL M3171 type I to type III. In that case, chromic acid or its salts are used for passivation. The

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use of sodium dichromate in combination with potassium permanganate has also been described (Dow Chemical Treatment, No. 22). Chemical passivation using aqueous passivating electrolytes containing chromium(VI) is simple to carry out. However, it has the serious disadvantage that the chromate substances, which are also present in the resulting conversion coatings, are carcinogenic.

Furthermore, the recyclability of chromated articles made of magnesium or its alloys represents a considerable problem, since because of their heavy metal content such articles require considerable effort in order to be recyclable to what are known as "high-purity" materials.

On grounds of environmental protection and workplace safety, a concern among manufacturers and processors of passivated articles made of magnesium or its alloys is to find a substitute for the conventional chromation, using chromate-free, aqueous passivating electrolytes.

Known chromate-free aqueous passivating electrolytes for the passivation of articles made from magnesium or its alloys are stannate-based aqueous passivating electrolytes which are marketed, for example, by the company Dow Chemical. However, it has been found that the corrosion protection effect of the resulting conversion coating is lower in comparison to the chromated magnesium materials.

US 5 743 971 describes a process for forming corrosion protection coatings on metals such as Zn, Ni, Ag, Fe, Cd, Al, Mg and their alloys.

In this process, these metals are immersed in a solution which comprises an oxidizing agent, a silicate and at least one cation from the group consisting of Ti, Zr, Ce, Sr, V, W and

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Mo. The pH of this solution is, in particular, in a range between 1.5 and 3.0.

The oxidizing agent is exclusively selected from the group of the peroxo compounds. A potassium permanganate oxidizing agent is not mentioned. Nor does this citation reveal the actual improvements provided by the process it describes for magnesium or its alloys in comparison to conventional chromations.

Moreover, the phosphation of articles made of magnesium or its alloys is also known (cf. Dow Chemical Treatment No. 18). Phosphating with simultaneous use of potassium permanganate is described in D. Hawk, D.L. Albright, "A Phosphate Permanganate Conversion Coating for Magnesium", Metal Finishing, October 1995, pp. 34 - 38. Here again, the corrosion protection obtained using these aqueous passivating electrolytes is significantly lower in comparison to a chromated coating.

A further possibility for chemical passivation is described by CHIBA Institute of Technology, Japan (published in the conference material INTERFINISHING 96 World Congress, Birmingham, UK, 10 - 12 September 1996, pp. 425 - 432), according to which a solution of potassium permanganate, alone or in combination with small amounts of acids ( $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{HF}$ ), is present in an aqueous passivating electrolyte. The aqueous passivating electrolyte temperature required for the chemical passivation is between 40 and 84 °C.

The conversion coating obtainable in this way exhibits a good protective effect; however, the stability of the aqueous passivating electrolyte is inadequate for an industrial application of this process. Thus, after a short time, magnesium dioxide ( $\text{MnO}_2$ ) is precipitated, rendering the aqueous passivating electrolyte unusable for the further passification of magnesium materials.

0934-0910  
101260-454550



An object of the invention is to provide a chemically passivated article made of magnesium or its alloys whose conversion coating is obtainable by an electrolytic, current-free process which is simple to apply and is transferable to the industrial scale. The corrosion protection effect of such a conversion coating, moreover, should not be poorer than that of the known, chromated articles made of magnesium or its alloys.

This object is achieved in accordance with the invention by means of an article made of magnesium or its alloys, some or all of whose surface has a conversion coating, characterized in that the conversion coating comprises  $MgO$ ,  $Mn_2O_3$  and  $MnO_2$  plus at least one oxide from the group consisting of vanadium, molybdenum and tungsten.

The conversion coating of the invention may be obtainable by passivating the article using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.

The object on which the invention is based is likewise achieved by means of a process for producing a conversion coating on an article made of magnesium or its alloys, characterized in that the article is subjected to passivation using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.

The conversion coating of the invention has a golden brown to greyish brown, iridescent colour and comprises  $MgO$ ,  $Mn_2O_3$ ,  $MnO_2$  and at least one oxide from the group consisting of vanadium, molybdenum and tungsten.

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Investigations have shown that the corrosion protection effect of this conversion coating is no lower than that of a conventional chromate coating.

Especially in view of the fact that, in comparison with the chromate ions, the anions used in accordance with the invention have when considered individually a lower oxidizing power than chromate ions, it is clear that only through combination of the permanganate ions with the corresponding vanadate, molybdate and/or tungstate ions is a synergistic effect achieved which leads to the formation of a corrosion-inhibiting conversion coating on articles made of magnesium or its alloys.

This is of particular significance since the prior art aqueous passivating electrolytes comprising potassium permanganate can only achieve such an oxidizing power of the electrolyte solution by a reduction in pH and/or increase in temperature.

One possible explanation for this synergistic effect may lie in the formation of very strong, so-called heteropolyacids in the form of their soluble ammonium or alkali metal salts.

A particular advantage of the process of the invention is the fact that, even after a relatively long time in use, the aqueous passivating electrolyte remains stable without the precipitation of magnesium dioxide in an amount which would render the aqueous passivating electrolyte unusable for the passivation of articles made of magnesium or its alloys.

Accordingly, it is possible with the present process in a simple manner simply to top up the chemicals which have been consumed following a prolonged time in use without the need to change the aqueous passivating electrolyte itself.

In accordance with one preferred embodiment of the present invention, a polymer coating obtainable by polymerizing and/or

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crosslinking a solution comprising at least one alkoxy silane compound is additionally applied to the conversion coating.

In this way, the mechanical and chemical properties of the conversion coating (e.g. corrosion resistance or operation resistance) are greatly increased. In this case, the conversion coating of the invention acts as an adhesion primer.

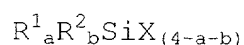
Accordingly, the conversion coating obtainable in accordance with the process of the invention has pores with a size of between 200 and 1000 nm.

The choice of an alkoxy silane compound as the compound to be polymerized and/or crosslinked ensures that the polymer coating present on the conversion coating is connected firstly as a result of chemisorption via Si-O bonds to the surface of the conversion coat and secondly by chemisorption inside the pores. The penetration of the alkoxy silane compound into the pores of the conversion coating brings about an increase in the contact area and thus in the chemisorption between conversion coating and polymer coating.

The polymer coating is formed by conventional polymerization processes familiar to the person skilled in the art (e.g. air drying, heating or UV radiation).

The amount of alkoxy silane compound in the solution to be applied may vary within wide limits. In general, the solution contains from 5 to 45% by weight, in particular from 10 to 30% by weight, of the alkoxy silane compound. Depending on the required viscosity, the solution may additionally comprise a polar solvent, which is to be chosen such that it does not react with the alkoxy silane compound (e.g. ethanol).

In accordance with one preferred embodiment, the alkoxy silane compound is of the general formula



in which

- X is an alkoxy, aryloxy or acyloxy group of 1 to 12 carbon atoms, preferably of 1 to 4 carbon atoms, and in particular is selected from the group consisting of methoxy, ethoxy, n-propoxy, i-propoxy, butoxy, phenoxy, acetoxy and propionyloxy groups;
- $R^1$  and  $R^2$ , which are identical to or different from one another, are selected from the group consisting of
  - amino, monoalkylamino or dialkylamino radicals;
  - alkyl radicals, especially the alkyl radicals of 1 to 6 carbon atoms, preferably the methyl, ethyl, n-propyl, isopropyl, n-butyl, s-butyl, t-butyl, pentyl, hexyl or cyclohexyl radicals;
  - alkenyl radicals, especially the alkenyl radicals of 2 to 6 carbon atoms, preferably the vinyl, 1-propenyl, 2-propenyl or butenyl radicals;
  - alkynyl radicals, especially the alkynyl radicals of 2 to 6 carbon atoms, preferably the acetylenyl or propargyl radicals;
  - aryl radicals, especially the aryl radicals of 6 to 10 carbon atoms, preferably phenyl or naphthyl radicals;
  - epoxy radicals, especially the epoxy radicals of 3 to 16 carbon atoms, preferably the glycidyl, glycidyl ether, glycidyl ester or glycidyloxyalkyl radicals; or
  - group X described above; and
- a and b, which are identical to or different from one another, are 0, 1, 2 or 3, the sum of a and b not exceeding 3.

Such an alkoxysilane compound may be a tetraalkoxysilane, epoxyalkoxysilane or aminoalkoxysilane.

Very good results have been obtained with tetraethoxysilane, 3-glycidyloxypropyltrimethoxysilane, 3-aminopropyltrimethoxysilane and 3-

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(aminoethylamino)propyltrimethoxysilane as the alkoxysilane compound.

In order to improve further the adhesion between conversion coating and polymer coating, it is advisable additionally to add a compound capable of forming a titanium complex to the solution that is to be applied to the conversion coating. The term "compound capable of forming a titanium complex" denotes compounds which form bridged  $\text{TiO}_2\text{-SiO}_2$  systems by complex-type bonding with the alkoxysilane compound and the conversion coating. Furthermore, the reaction between alkoxysilane compound and titanium compound produces a crosslinked polymer coating.

A particularly suitable compound is an alkoxytitanium compound, a titanate ester or a titanium chelate, especially a compound of the formula  $\text{Ti(OR)}_4$ , in which R is an alkyl radical of 1 to 6 carbon atoms which is preferably selected from the group consisting of methyl, ethyl, n-propyl, i-propyl and butyl radicals.

Very good results have been achieved with tetraethoxytitanate  $\text{Ti(OC}_2\text{H}_5)_4$ .

The molar ratio between alkoxysilane compound and titanium compound is not critical and is generally between 1 and 20.

Solutions which include both an alkoxysilane compound and a compound capable of forming a titanium complex are described, for example, in DE 41 38 218 A1 and may be purchased from various companies (e.g.. Deltacoll® 80 from Dörken).

Where necessary, the polymer coating may also have a colour. In this case, the solution to be polymerized and/or crosslinked additionally comprises at least one dye which is soluble in a polar solvent, in particular a metal complex dye. A metal complex dye of this kind is available, for example,

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under the commercial designation Neozapon<sup>®</sup> from BASF, Orasol<sup>®</sup> from Ciba-Geigy, Savinyl<sup>®</sup> from Sandoz or Lampronol<sup>®</sup> from ICI. Owing to the solubility of the dye in a polar solvent, a homogeneous solution and, accordingly, a homogeneous structure of the polymer coating is achieved. There is therefore no accumulation of the dye in the polymer coating, which might otherwise act as a "built-in" fracture point between conversion coating and polymer coating.

In the process of the invention for producing a conversion coat, the passivation is preferentially conducted within a pH range of the aqueous passivating electrolyte of from 7.0 to 8.0.

Accordingly, it is possible to forego the addition of acids. This means that it is unnecessary to reduce the pH by adding acids in order to increase the oxidizing power of the permanganate anions.

Furthermore, it is possible for the first time with the process of the invention to conduct a sufficient passivation at a temperature of the aqueous passivating electrolyte of from 15 to 50 °C, in particular from 20 to 30 °C.

The passivation is usually conducted for a period of from 2 to 10 minutes.

The concentration of potassium permanganate in the aqueous passivating electrolyte of the invention is preferably from 1 to 10 g/l; that of the alkali metal or ammonium salt of the vanadate, molybdate and/or tungstate ions is preferably from 1 to 10 g/l. In particular, the upper limit on the vanadate, molybdate and/or tungstate concentration is not critical. Thus the process of the invention can be conducted even with an electrolyte which comprises a saturated solution of these salts, even with undissolved constituents.

The synergistic effect between permanganate ions and vanadate, molybdate and/or tungstate ions becomes particularly clear when it is attempted to passivate an article made of magnesium using only an aqueous potassium permanganate solution with a concentration of from 1 to 10 g/l, with identical operating parameters. This is because, under these conditions, it is not possible to obtain a conversion coating affording an adequate corrosion protection effect.

The articles passivated in accordance with the invention comprise, for example, parts for the motor vehicle industry, electrical and electronics industry, mechanical engineering industry, the aerospace industry, and parts of sports equipment.

Particular mention may be made of parts of engines and transmission casings, instrument panels, doors and individual parts of them, steering gear casings, wheel spiders for motorbikes, throttle valve casings, receptacles for milling cutters, rotors or displacer casings for compressors, sealing jaws for packaging machines, parts for plug strips and electrical connectors, lamp bases, lamp housings, rotor casings of helicopters, housings for electrical equipment, and parts of sporting bows.

Magnesium alloys which may be employed in particular are all customary diecasting, casting and wrought alloys. Examples are, in particular, AZ91, AZ81, AZ61, AM60, AM50, AM20, AS41, AS21, AE42, QE22, ZE41, ZK61 and AZ31, AZ60, ZK30, ZK60, WE43 and WE54 (ASTM codes).

The invention further relates to the use of a solution for producing an article of the invention, this solution comprising at least one of the alkoxysilane compounds described above.

Furthermore, it is possible to apply a paint or other coating material in addition to the conversion coating, with or without an additional polymer coating.

The following examples serve to illustrate the invention.

12 sheets of the magnesium alloy AZ91HP measuring 50 x 100 x 2 mm are chromated in accordance with the MIL specification M3171 type I.

The sealant used is a silane combination (DELTACOLL 80 from Dörken) and/or an epoxy-polyester powder coating material (Delta-S-NT-Pulverlack from Dorken) in accordance with the conditions specified in Table I.

The results of the salt spray test are reported in Table I.



### Example 1

12 sheets of the magnesium alloy AZ91HP measuring 50 x 100 x 2 mm are pickled in 75% strength  $H_3PO_4$  for 30 seconds. They are then rinsed with deionized water and neutralized in 10% strength NaOH at room temperature for 30 seconds; thereafter, the sheets are again rinsed with deionized water. In the wet state, the sheets are immersed in an aqueous passivating electrolyte at room temperature for 5 minutes, the electrolyte consisting of an aqueous solution of 3 g/l  $KMnO_4$  and 1 g/l  $NH_4VO_3$ . Following the removal of the sheets from the passivating bath, the conversion coating, which has a grey-brown appearance, is rinsed with deionized water and then dried at 110 °C for 30 minutes.

Batches of three of the sheets passivated in this way are subjected, in the original condition (without sealing), and sealed with special coatings, to a salt spray test in accordance with DIN 50021-SS.

The sealant used is a silane combination (DELTACOLL 80 from Dörken) and/or an epoxy-polyester powder coating material (Delta-S-NT-Pulverlack from Dörken) in accordance with the conditions specified in Table I.

The results of the salt spray test are reported in Table I.

DELTA-S-NT-PULVERLACK

Table I

	Comparative Example 1 [h]*	Example 1 [h]*
Passivation without sealing	5 - 10	5 - 10
Passivation + silane combination (DELTACOLL 80)	412 - 495	451 - 608
Passivation + epoxy polyester coating material (Delta- Pulverlack) 80 to 100 µm	505 - 603	528 - 607
Passivation + silane combination (DELTACOLL 80) + epoxy polyester powder coating (Delta-Pulverlack) 80 to 100 µm	796 - 1038	818 - 1038

\*: The smaller figure corresponds to the time at which the first of the three sheets shows signs of inadequate corrosion protection; the larger figure indicates the time at which the last of the three sheets shows signs of inadequate corrosion protection.

#### Comparative Example 2

6 sheets of the magnesium alloy AM50HP measuring 50 x 100 x 2 mm are chromated in accordance with the MIL specification M3171 type I.

Batches of three of the sheets passivated in this way are subjected, in the original condition (without sealing), and sealed with a silane combination (DELTACOLL 80 from Dörken), to a salt spray test in accordance with DIN 50021-SS.

The results of the salt spray test are reported in Table II.

#### Example 2

6 sheets of the magnesium alloy AM50HP measuring 50 x 100 x 2 mm are pickled in 40% strength HF at room temperature for 60

DELTA 80 + EP 80

seconds. After rinsing with deionized water, the sheets are immersed in an aqueous passivating electrolyte consisting of an aqueous solution containing 4 g/l  $\text{KMnO}_4$  and 1.5 g/l  $\text{Na}_2\text{WO}_4$  at room temperature for 10 minutes. Following the removal of the sheets, the conversion coating, which has a golden-brown iridescence, is rinsed with deionized water and dried at 110 °C for 60 minutes.

Batches of three of the sheets passivated in this way are subjected, in the original condition (without sealing), and sealed with a silane combination (DELTACOLL 80 from Dörken), to a salt spray test in accordance with DIN 50021-SS.

The results of the salt spray test are reported in Table II.

Table II

	Comparative Example 2 [h]*	Example 2 [h]*
Passivation without sealing	5 - 10	5 - 10
Passivation + silane combination (DELTACOLL 80)	483 - 694	552 - 745

\*: The smaller figure corresponds to the time at which the first of the three sheets shows signs of inadequate corrosion protection; the larger figure indicates the time at which the last of the three sheets shows signs of inadequate corrosion protection.

### Comparative Example 3

6 sheets of magnesium alloy AZ91HP measuring 50 x 100 x 2 mm are chromated in accordance with the MIL specification M3171 type I.

Batches of three of the sheets passivated in this way are sealed with a silane combination (DELTACOLL 80 from Dörken) and with an epoxy-polyester powder coating material (Delta-S-NT-Pulverlack from Dörken) and then subjected to a salt spray test in accordance with DIN 50021-SS.

The number of points of corrosion was determined as a function of time. The results are reported in Table III.

### Example 3

6 sheets of AZ91HP measuring 50 x 100 x 2 mm are pickled in 75 % strength  $H_3PO_4$  for 30 seconds. They are then rinsed with deionized water, neutralized with 10% strength aqueous NaOH for 45 seconds, and then rinsed with deionized water again. Then, in the wet state, the sheets are immersed in an aqueous passivating electrolyte consisting of an aqueous solution of 3 g/l  $KMnO_4$  and 1 g/l  $NaVO_3$  at room temperature for 4 minutes. Following the removal of the sheets, the conversion coating, which has a grey-brown appearance, is rinsed with deionized water and then dried at 110 °C for 45 minutes.

Batches of three of the sheets passivated in this way are sealed with a silane combination (DELTACOLL 80 from Dörken) and with an epoxy-polyester powder coating material (Delta-S-NT-Pulverlack from Dörken) and then subjected to a salt spray test in accordance with DIN 50021-SS.

The number of points of corrosion was determined as a function of time. The results are reported in Table III.

Table III

	Corrosion points after 100 h	Corrosion points after 200 h	Corrosion points after 350 h
Example 3 + silane combination (DELTACOLL 80)	0	0	1
Comparative Example 3 + silane combination (DELTACOLL 80)	3	4	8

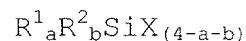
Example 3 + epoxy-polyester powder coating material (Delta- Pulverlack) 80 to 100 µm	0	0	0
Comparative 3 + epoxy-polyester powder coating material (Delta- Pulverlack) 80 to 100 µm	0	0	1

Table III clearly shows improved corrosion protection for the conversion coating of the invention when a silane combination is used.

[illegible]

## C l a i m s

1. Article made of magnesium or its alloys, some or all of whose surface has a conversion coating, characterized in that the conversion coating comprises  $\text{MgO}$ ,  $\text{Mn}_2\text{O}_3$  and  $\text{MnO}_2$  plus at least one oxide from the group consisting of vanadium, molybdenum and tungsten.
2. Article according to Claim, 1 characterized in that the conversion coating is obtainable by passivating the article using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
3. Article according to Claim 1 or 2, characterized in that in addition to the conversion coating a polymer coating has been applied which is obtainable by polymerizing and/or crosslinking a solution comprising at least one alkoxy silane compound.
4. Article according to Claim 3, characterized in that the alkoxy silane compound is of the general formula



in which

- X is an alkoxy, aryloxy or acyloxy group of 1 to 12 carbon atoms, preferably of 1 to 4 carbon atoms, and in particular is selected from the group consisting of methoxy, ethoxy, n-propoxy, i-propoxy, butoxy, phenoxy, acetoxy and propionyloxy groups;

- $R^1$  and  $R^2$ , which are identical to or different from one another, are selected from the group consisting of
  - amino, monoalkylamino or dialkylamino radicals;
  - alkyl radicals, especially the alkyl radicals of 1 to 6 carbon atoms, preferably the methyl, ethyl, n-propyl, isopropyl, n-butyl, s-butyl, t-butyl, pentyl, hexyl or cyclohexyl radicals;
  - alkenyl radicals, especially the alkenyl radicals of 2 to 6 carbon atoms, preferably the vinyl, 1-propenyl, 2-propenyl or butenyl radicals;
  - alkynyl radicals, especially the alkynyl radicals of 2 to 6 carbon atoms, preferably the acetylenyl or propargyl radicals;
  - aryl radicals, especially the aryl radicals of 6 to 10 carbon atoms, preferably phenyl or naphthyl radicals;
  - epoxy radicals, especially the epoxy radicals of 3 to 16 carbon atoms, preferably the glycidyl, glycidyl ether, glycidyl ester or glycidyloxyalkyl radicals; or
  - group X described above; and
- a and b, which are identical to or different from one another, are 0, 1, 2 or 3, the sum of a and b not exceeding 3.

5. Article according to Claim 4, characterized in that the alkoxyasilane compound is a tetraalkoxyasilane, epoxyalkoxyasilane or aminoalkoxyasilane.
6. Article according to Claim 5, characterized in that the alkoxyasilane compound is selected from the group consisting of tetraethoxyasilane, 3-glycidyloxypropyltrimethoxyasilane, 3-aminopropyltrimethoxyasilane and 3-(aminoethylamino)propyltrimethoxyasilane.
7. Article according to any of Claims 3 to 6, characterized in that the solution additionally comprises a compound capable of forming a titanium complex.

8. Article according to Claim 7, characterized in that the compound capable of forming a titanium complex is an alkoxytitanium compound, a titanate ester or a titanium chelate and in particular is of the formula  $Ti(OR)_4$  in which R is an alkyl radical of 1 to 6 carbon atoms selected preferably from the group consisting of methyl, ethyl, n-propyl, i-propyl and butyl radicals.
9. Article according to Claim 8, characterized in that the compound capable of forming a titanium complex is tetraethoxytitanate  $Ti(OC_2H_5)_4$ .
10. Article according to any of Claims 3 to 9, characterized in that the solution additionally comprises at least one dye which is soluble in a polar solvent, in particular a metal complex dye.
11. Process for producing a conversion coating on an article made of magnesium or its alloys, characterized in that the article is subjected to passivation using an aqueous passivating electrolyte which comprises potassium permanganate and at least one alkali metal salt or ammonium salt of an anion from the group consisting of vanadate, molybdate and tungstate.
12. Process according to Claim 11, characterized in that the passivation is conducted within a pH range of the aqueous passivating electrolyte of from 7.0 to 8.0.
13. Process according to Claim 11 or 12, characterized in that the passivation is conducted at a temperature of the aqueous passivating electrolyte of from 15 to 50 °C, in particular from 20 to 30 °C.



14. Process according to any of Claims 11 to 13, characterized in that the passivation is conducted for a period of from 2 to 10 minutes.
15. Process according to any of Claims 11 to 14, characterized in that the concentration of potassium permanganate in the aqueous passivating electrolyte is from 1 to 10 g/l.
16. Process according to any of Claims 11 to 15, characterized in that the concentration of the alkali metal salt or ammonium salt from the group consisting of vanadate, molybdate and tungstate in the aqueous passivating electrolyte is from 1 to 10 g/l.
17. Process according to any of Claims 11 to 16, characterized in that a paint or other surface coating material is or has been applied to the conversion coating.
18. Use of a solution for producing an article according to any of Claims 3 to 10, characterized in that the solution comprises at least one alkoxysilane compound according to any of Claims 3 to 9.
19. Use of an article according to any of Claims 1 to 10 and of an article obtainable by a process as claimed in any of Claims 11 to 17 in the motor vehicle industry, electrical and electronics industry, mechanical engineering industry, air travel and space travel.

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PATENT APPLICATION  
(37 CFR 1.63)**

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OR  
☐ Declaration Submitted after Initial Filing (surcharge (37 CFR 1.16 (e)) required)

Attorney Docket Number

First Named Inventor

COMPLETE IF KNOWN

Application Number

Filing Date

Group Art Unit

Examiner Name

As a below named inventor, I hereby declare that:

My residence, mailing address, and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

CHEMICALLY PASSIVATED OBJECT MADE OF MAGNESIUM  
OR ALLOYS THEREOF

(Title of the Invention)

the specification of which

☐ is attached hereto

OR

☒ was filed on (MM/DD/YYYY) 03/22/2000 as United States Application Number or PCT International

Application Number PCT/DE00/000872 and was amended on (MM/DD/YYYY) 10/06/2000 (if applicable).

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment specifically referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR 1.56, including for continuation-in-part applications, material information which became available between the filing date of the prior application and the national or PCT international filing date of the continuation-in-part application.

I hereby claim foreign priority benefits under 35 U.S.C. 119(a)-(d) or (f), or 365(b) of any foreign application(s) for patent, inventor's or plant breeder's rights certificate(s), or 365(a) of any PCT international application which designated at least one country other than the United States of America, listed below and have also identified below, by checking the box, any foreign application for patent, inventor's or plant breeder's rights certificate(s), or any PCT international application having a filing date before that of the application on which priority is claimed.

Prior Foreign Application Number(s)	Country	Foreign Filing Date (MM/DD/YYYY)	Priority Not Claimed	Certified Copy Attached?	
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NAME OF SOLE OR FIRST INVENTOR: ☐ A petition has been filed for this unsigned inventorGiven Name  
(first and middle [if any]) PeterFamily Name  
or Surname KurzeInventor's  
Signature Peter KurzeDate 5 April 2001Residence: City Nideggen

State

Germany  
CountryGerman  
CitizenshipMailing Address Lagerstrasse 10, 52385City Nideggen

State

ZIP

Country GermanyNAME OF SECOND INVENTOR: ☐ A petition has been filed for this unsigned inventorGiven Name  
(first and middle [if any]) UlrikeFamily Name  
or Surname KrugerInventor's  
Signature U.K.Date 5 April 2001Residence: City Kerpen

State

Germany  
CountryGerman  
CitizenshipMailing Address Falladastrasse 12, 50171City Kerpen

State

ZIP

Country Germany☐ Additional inventors are being named on the 1 supplemental Additional Inventor(s) sheet(s) PTO/SB/02A attached hereto

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## DECLARATION

## ADDITIONAL INVENTOR(S)

Supplemental Sheet

Page 3 of 3

Name of Additional Joint Inventor, if any:

☐ A petition has been filed for this unsigned inventor

Given Name (first and middle [if any])

Family Name or Surname

Marco

Kohler

Inventor's  
Signature

*Marco Kohler*

Date 5 April 2001

Residence: City Bornheim

State

Country Germany

Citizenship German

Mailing Address Hordorfer Weg 49, 53332

Mailing Address

City Bornheim

State

ZIP

Country Germany

Name of Additional Joint Inventor, if any:

☐ A petition has been filed for this unsigned inventor

Given Name (first and middle [if any])

Family Name or Surname

Dora

Banerjee

Inventor's  
Signature

*Dora Banerjee*

Date 5 April 2001

Residence: City Kerpen

State

Country Germany

Citizenship German

Mailing Address Regengasse 10, 50171

Mailing Address

City Kerpen

State

ZIP

Country Germany

Name of Additional Joint Inventor, if any:

☐ A petition has been filed for this unsigned inventor

Given Name (first and middle [if any])

Family Name or Surname

Inventor's  
Signature

Date

Residence: City

State

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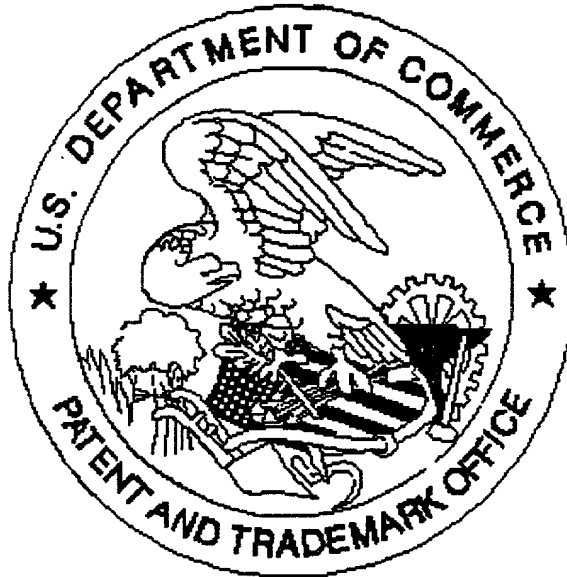
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